# Hydrochemical Investigation of Saline Water Intrusion into Aquifers in Part of Eastern Dahomey Basin, Southwestern Nigeria

Adeyemo Igbagbo Adedotun<sup>\*</sup>, Omosuyi Oluwole Gregory and Adelusi Adebowale Obasanmi

www.iiste.org

IISTE

Department of Applied Geophysics, Federal University of Technology, Akure, Nigeria

### ABSTRACT

This study is a major attempt at delineating presence and lateral extent of saline water intrusions into aquifers at the easternmost part of Dahomey basin which falls essentially in the sedimentary terrain of Ondo State of Nigeria. 61 water samples were collected from hand dug wells, shallow boreholes, and ponds across the study area and analyzed for relevant parameters such as pH, conductivity, total hardness, calcium hardness, magnesium hardness, total dissolved solids, alkalinity and concentrations of the following anions and cations; chloride, calcium, Sulphate, bicarbonate, magnesium and sodium. Equivalent salinity was calculated from the water sample analysis results. The hydrochemical analysis results reveals possible saline water intrusion in the coastal area, especially the southeastern part and Agbabu in the north central part of the study area as evident from high concentration values of chloride (372 - 1500 mg/l), alkalinity (105 - 330 mg/l), equivalent salinity (135 - 2808 mg/l), total dissolved solid (181 - 1005 mg/l), high pH values (4.4 - 8.6 pH) and conductivity values (541 - 1500 µs/cm).

Keywords: Saline water intrusion, saline-freshwater boundary, hydrochemical and equivalent salinity.

### 1. INTRODUCTION

The larger percentage of Nigerians relies entirely on groundwater abstraction for their domestic and industrial water needs. This heavy reliance on groundwater is perhaps due to the fact that it is available in all seasons and chemically stable. Most Nigerians rely on groundwater abstraction because it is cheaper to develop. Since Nigerians rely heavily on groundwater for various uses, it is important that groundwater potability be guaranteed at all times and therefore must be given serious attention. Groundwater quality is usually compromised, as a result of contamination or pollution deriving from anthropogenic activities, or as a result of saline water intrusion into coastal aquifers proximal to oceans across the world. Contamination of groundwater which could be either pollution or salinity can render groundwater resources useless, unwholesome and unfit for consumption and other purposes.

Saline water intrusion can simply be defined as movement of saline water or sea water into fresh water aquifer or surface reservoir. When the source of the saline water is sea water, then the phenomenon can be referred to as seawater intrusion (Rahaman and Bhattacharya, 2014). Ghyben (1888) and Herzberg (1901) described what happens at fresh-saline water boundary. The fresh-saline water boundary is not sharp but gradational and water within this transition zone is known as brackish water. Since saline water is denser because of its higher concentration of dissolved salt, it will normally remain below fresh water and will only intrude into the fresh water when hydraulic pressure within the saline water is higher than that of the overlying fresh water. Fresh-saline water boundary has also been observed to shift seaward during rainy season, because of strong pressure of fresh water when been recharged by precipitation, while the reversal is the case during dry season. It

has also been observed that good drainage system reduces soil salinity (Mohammad, 2014).

The possible sad consequence of consuming saline or brackish water came to light last year when some Nigerians consumed salt solution supposedly to enhance their body immune system against the deadly Ebola virus, few people lost their lives across the country as a result of complications arising from high concentration of salt in the consumed salt solution. This incident clearly showed that consumption of saline or brackish water is dangerous to human health.

Only few works have been done on saline water intrusion in the study area. Omosuyi et. al. (2008) carried out geoelectric sounding, hydrogeochemical and hydrogeophysical studies of the eastern Dahomey basin, and the adjoining basement complex in southern part of Ondo state. Fifty (50) Schlumberger vertical electrical sounding (VES) was carried out across southern part of Ondo State, using maximum current electrode separation (AB/2) of 325 m. From water analysis result it was found out that the TDS in most water samples analyzed are within the World Health Organization acceptable limit with the exception of Arogbo and Igbokoda which are higher. Omoyoloye et al. (2008) carried out 16 VES across Adagbakuja Newtown in southern part of Ondo state, using maximum current electrode separation (AB/2) of 130 - 225 m. Four (4) geoelectric layers were delineated; topsoil (0.4 - 199 ohm-m), mud and peat (0.4 - 102 ohm-m), clayey silts/fine sand (0.4 - 76 ohm-m) and brackish/saline water and sand units (2 - 1528 ohm-m). Brackish or saline water intrusion was delineated within the silt/fine sand substratum. This present work was aimed at delineating presence and lateral extent of saline water intrusion in the study area.

### 2. DECSRIPTION OF THE STUDY AREA

The study area is generally characterized by flat and gently undulating topography. Topographic elevations vary from about 13 to 83 m above sea level in the mainland and 2 to 10 m in the coastal area (Figure 1). Most part of the coastal area is water logged, which makes the area vulnerable to sea water incursion. The study area is been drain by many tributaries, streams and rivers some of which are connected directly or indirectly to the Ocean and this can facilitates the rate and extent of saline water intrusion. The area cuts across five Local Government areas in the southern part of Ondo State (Odigbo, Irele, Okitipupa, Ese-Odo and Ilaje) and Ogun Waterside Local Government in Ogun State. It is bounded by the following coordinates; longitudes and latitudes,  $4^{\circ} 22^{'} 22.5^{''}$ E and  $5^{\circ} 10^{'} 2.0^{''}$ E and  $5^{\circ} 50^{'} 44.1^{''}$ N and  $6^{0} 39^{'} 39.5^{''}$ N and it cover a total area of about 4,200 km<sup>2</sup>.

#### 3. GEOLOGY OF THE STUDY AREA

The study area lies within the easternmost part of Dahomey basin, southwestern Nigeria. It is bordered to the north by the crystalline rocks of the southwestern Nigeria Basement Complex and to the South by Atlantic Ocean christened the Bight of Benin (Figure 2). The Dahomey Basin is a combination of inland/coastal/offshore basin that stretches from south-eastern Ghana through Togo and the Republic of Benin to south-western Nigeria. It is separated from the Niger Delta by a subsurface basement high referred to as the Okitipupa Ridge. Its offshore extent is poorly defined (Billman, 1992). The Dahomey Basin covers much of the continental margin of the Gulf of Guinea, extending from Volta-Delta in Ghana in the west to the Okitipupa Ridge in Nigeria in the east. The basin is a marginal pull-apart basin (Klemme, 1975) which developed in the Mesozoic due to the separation of African from Southern American plate in the Mesozoic era (Burke et al; 1971; Whiteman, 1982). The eastern Dahomey basin or the Nigeria sector contains extensive wedge of Cretaceous to Recent sediments, up to 3000 m which thicken towards the offshore. The sediment deposition in the basin follows an east-west trend. The

summary of the general stratigraphy succession in the Nigeria section of the eastern Dahomey Basin which covers Ogun State and the southern part of Ondo State and south western part of Edo State of Nigeria is as presented in table 1.

### 4. METHODOLOGY

A total 61 water samples (Figure 3) were collected across the study area during a period that spans through the dry and wet seasons, from January, 2012 to August 2013. Most of the water samples were obtained from hand dug wells with depth range of 3 to 6 m. Borehole water was obtained at 12 locations which are mainly in the southeastern part of the study area; they are Agbabu, Irele, Okitipupa, Molutehin, Akpovukoko, Odonla, Awoye, Rewoye, Obenla, Ayetoro, Idogun, and Eruna. Surface water (pond) was obtained in 5 locations; Orioke, Asisa, Ugbonla road, Orere-ara road and Ileriayo. 17 primary parameters were determined from the hydrochemical analysis and they includes some of the following; conductivity, pH value, total hardness, calcium hardness, magnesium hardness, sodium, total dissolved solids (TDS), alkalinity, chloride, sulphate, bicarbonate, pH and conductivity. Equivalent salinity was derived from some of the primary hydrochemical parameters. Conductivity and pH were determined insitu using a hand held devices, while other parameters were determined in the laboratory.



Figure 1: Topographic map of the study area



Figure 2: Geological map of the study area (after PTF, 1997)

Group	Formation	Lithology	Age
	Benin	Poorly sorted sands with lenses of clay	Oligocene-recent
	Ilaro	Massive, yellowish poorly consolidated cross-bedded sandstones	Eocene
	Oshosun	Greenish or Beige clay with interbeds of sandstones	Eocene
	Akinbo	Shale and Clayey sequence	Palaeocene-Eocene
	Ewekoro	Limestone with presence of Coralline algae, gastropods, pelecypods, echinoids fossils	Palaeocene
Abeokuta	Araromi	Shales, siltstone with interbedded limestone, marl and lignite	Maastritchian-Palaeocene
		Medium-fine grained sandstone.	
	Afowo	Coarse-medium grained sandstone; Interbedding of shales, siltstones and claystones	Turonian-Maastritchian
	Ise	Coarse-medium grained sands	Neocomian-Albian
		Conglomerates and Grits	
	<u> </u>	Basement Complex	Precambrian

## Table 1: Summary of the Geology of the study area

### 5. DISCUSSION OF RESULTS

The results of the water analyses were as presented in table 2 and various maps. Although pH usually has no direct impact on consumers, it is one of the most important operational water quality parameters, the optimum required pH values range from 6.5 - 9.5 (WHO, 2008). The pH map (Figure 4) shows that the pH values across the study area range from 4.4 (Asisa) to 8.6 (Araromi Seaside), which is still within the WHO acceptable limit (Table 3). However places like Araromi Obu, Lomiroro and Aiyesan in the northwestern part, Agbabu and Ilubirin in the northcental part and the coastal towns of the study area exhibit high pH values indicating that alkalinity increase towards the coast.

Chloride concentration across the study area varies from 4.28 (Akotogbo) to 1500 mg/l (Orioke and Asisa). The Chloride map (Figure 5) shows that the chloride concentration is high at north central (Agbabu) and southeastern parts (Mahin, Ugbo, Ugbonla, Idogun,) of the study area. The WHO maximum permissible value for chloride concentration is 250 mg/l (Table 3). This value was exceeded at Ugbonla, Idogun, Orioeke, Asisa and Agbabu, where chloride concentration values of 372 to1500 mg/l were obtained.



Figure 3: Location map of the study area showing water sampling locations

The alkalinity values derived from all the water samples range from of 4 - 330 mg/l. The minimum value of 4 mg/l was obtained from samples collected from Efire and Erinje, while high alkalinity values above 100 mg/l, which are higher than the WHO maximum permissible limit (Table 3), were obtained from the following locations Mahin and Ugbonla (104 mg/l), Orioke (103 mg/l), Rewoye (120 mg/l), Obenla (130 mg/l), Odonla (166 mg/l), Igbokoda (184 mg/l), Ugbo (196 mg/l) and Agbabu (200 mg/l), Awoye, (205 mg/l), Eruna (258 mg/l), Molutehin (260 mg/l) and Ayetoro (305 mg/l). The alkalinity map (Figure 6) shows that alkalinity is high around

the southeastern part of the study area, where Mahin, Ugbo, Ugbonla, Awoye, Molutehin, Obenla, Odonla, Idogun and Ayetoro are situated. Also another high alkalinity area is the north central (or the mainland) part of the study area around Agbabu.

Equivalent salinity values were derived from Sodium (Na<sup>+</sup>), Chloride (Cl<sup>-</sup>), Calcium (Ca<sup>+2</sup>) and Sulphate (So<sub>4</sub><sup>-2</sup>) concentrations, based on the following relationship;

Equivalent Salinity Concentration =  $0.81Ca + 0.45SO_4 + (Na+Cl)$ 

The equivalent salinity values range from 25.8 (Ode-Irele) to 2808 mg/l (Asisa). The equivalent salinity map (Figure 7) shows that high values were obtained from Ugbo, Ugbonla, Odonla, Idogun and Ayetoro area in the southeastern part of the study area and likewise at Agbabu in the north central part of the study area. Though there is no WHO standard for equivalent salinity, however the map correlates well with alkalinity map (Figure 6). These showed that shallow and intermediate aquifers in these areas are yielding saline or brackish water.

Total dissolved solid (TDS) is a useful parameter commonly used in determining saline water intrusion and has been used by many authors (Omosuyi, 2001; Choudhury et al., 2001 and Samsudin et al., 2007). The TDS values across the study area (Figure 8) range from 6.7 to 1005 mg/l. The lowest value of 6.7 mg/l was recorded from OdeIrele, Ajagba, Iju-Osun Akpovukoko, Akotogbo, Agadagba, Igbekebo, Iju-Odo, Efire and Ilusin. High TDS values were obtained at Ugbonla (582.9 mg/l), Ugbo (670 mg/l), Orioke (836 mg/l), Asisa (970 mg/l) and Agbabu (1005 mg/l). In all of these locations, the water samples analysis yielded TDS values well above 500 mg/l which appears to be indicative of saline water intrusion in the study area, however the maximum permissible value of 1500 mg/l is recommended by WHO (2008).

Conductivity has a direct relationship with salinity; increases in salinity will lead to increase in the concentration of current carrying ions in water and consequently increases in conductivity. Therefore an increase in salinity, under a normal condition will corresponds to increase in conductivity, and therefore high conductivity is strongly connected to saline water intrusion. The maximum permissible limit of conductivity is 1200 µs/cm (WHO, 2008). The conductivity map (Figure 9) shows that water samples collected from the northeastern (Odeaye, Ilubirin and Agbabu) and the southeastern (Igbokoda, Ugbo, Awoye, Ugbonla, Idogun, Asisa, Orioke and Ayetoro) parts of the study area yielded relatively higher conductivity values. However in all the samples analyzed, conductivity was still within WHO acceptable limits.

### 6. CONCLUSION

The hydrochemical analysis results as presented from various maps shows evidence of saline water intrusion in the southeastern part of the coastal and Agbabu in the mainland as evident from high concentration values of chloride (372 - 1500 mg/l), alkalinity (105 - 330 mg/l), equivalent salinity (135 - 2808 mg/l), total dissolved solid (181 - 1005 mg/l), high values of pH (4.4 - 8.6 pH) and conductivity (541 - 1500 µs/cm). In all of these results Orioke and Asisa have the highest values followed by other locations in coastal area and surprisingly Agbabu in the main land. However there is possibility that the whole southern part of the study area have been intruded by saline water. All the sampling points in the southwestern area are either shallow hand dug well (with depth less than 6 m) or surface water; this is due to non-availability of functioning motorized boreholes in this area, while the opposite is the case in the southeastern part of the

30	29	28	27	26	25	24	23	22	21	20	19	18	17	16	15	14	13	12	11	10	9	8	7	6	5	4	3	2	1	S/N
Araromi-Obu	Makun	Atijere	Efire	Lomiroro	Aiyesan	Aboto	Ugbonla	Ugbo	Mahin	Igbokoda	Okitipupa	Igbotako	Iju-Odo	Ilutitun	Ikoya	Ode-Aye	Igbekebo	Amapere	Agadagba	Iju-Osun	Akotogbo	Akpovukoko	Ajagba	Sabomi	Ugbonla Road	Olu-Agbo	Igbobini	Ode-Irele	Ayadi	Location Name
60	200	90	10	160	70	60	870	1000	270	600	80	60	10	40	50	680	10	50	10	10	10	10	10	110	666	30	50	10	50	Conductivity
8.23	7.74	7.82	6.93	5.6	7.21	6.01	7.15	7.3	7.11	6.89	4.45	4.51	5.05	5.49	4.55	5.88	4.99	5.21	5.14	4.92	4.87	4.95	5.23	6.32	6.73	5.77	6.08	5.45	5.9	рH
37	94	86	30	38	34	28	438	368	136	180	32	18	34	18	22	118	20	20	18	18	20	18	20	74	54	18	34	18	46	Total Hardness
18	72	54	20	14	16	14	302	232	110	143	14	10	20	12	10	92	10	14	8	10	10	12	10	64	14	8	22	10	18	Calcium Hardness
19	22	44	10	20	18	14	132	133	26	37	18	8	14	6	12	26	10	6	10	8	10	6	10	10	40	10	12	8	28	Total Hardness Calcium Hardness Magnesium Hardness
1.28	ND	4.4	4.84	3.24	8.36	1.2	1.24	1.08	1.76	0.25	0.48	0.86	0.78	0.64	1.11	0.74	0.68	0.47	0.97	0.86	0.81	0.74	0.63	0.48	0.47	0.83	11.44	0.88	0.97	Nitrate
12	24	17	9.99	26	11	15.7	372	132.8	17.9	59.3	10.7	14.3	5.71	7.14	9.28	67.8	4.28	12.9	4.99	4.99	4.28	7.86	5.71	5.71	13	8.57	9.99	8.57	12.9	Chloride
0.01	0.008	0.016	0.013	ND	0.012	0.008	ND	0.012	ND	0.015	ND	ND	ND	0.009	0.018	ND	0.015	0.017	0.011	0.015	0.02	0.021	0.015	0.014	0.003	0.017	0.012	0.018	0.015	Chloride Manganese
14	20	12	4	8	6	24	104	196	104	184	16	12	12	10	16	20	22	14	9	10	16	20	8	28	60	20	14	12		ese Alkalinity
32.7	41.9	25.1	43.2	28.6	34.2	26.7	19.8	46.4	28.7	63.4	52.4	17.9	36.2	24.3	16.9	28.4	64.3	42.4	44.6	56.5	43.4	18.9	38.3	42.3	1	32.4	26.4	18.7	33.4	y Sulpahte
0.61	0.69	0.93	0.61	0.62	0.96	1.2	ND	0.14	0.32	0.67	ND	ND	ND	0.35	0.28	ND	0.22	0.01	0.09	0.73	0.61	0.06	0.3	0.28	0.06	ND	0.08	0.42	0.73	Iron
7.8	15.6	11.1	6.49	16.9	7.15	10.2	241.8	86.3	11.6	38.5	6.96	9.3	3.71	4.64	6.03	44.1	2.78	8.39	3.24	3.24	2.78	5.11	3.71	3.71	8.45	5.57	6.49	5.57	8.39	Sodium
14	20	12	4	8	6	24	104	196	104	184	16	12	12	10	16	20	22	14	10	10	16	20		28			14	_	_	Bicarbonate
7.21	28.9	21.6	8.02	5.61	6.41	5.6	121	92.9	44.1	57.3	5.6	4	8	4.8	4	36.9	4	5.6	3.2	4	4	4.8	4	25.7	5.61	3.2	8.8	4	7.2	Calcium
4.64	5.37	10.7	2.44	4.88	4.39	3.4	32.2	32.5	6.3	9	4.4	1.9	3.4	1.5	2.9	6.3	2.4	1.46	2.44	1.95	2.44	1.46	2.44	2.44	9.76	2.44	2.92	1.95	6.8	Magnesium
40.2	134	60.3	6.7	107	46.9	40.2	582.9	670	180.9	402	53.6	40.2	6.7	26.8	33.5	455.6	6.7	33.5	6.7	6.7	6.7	6.7	6.7	73.7	446	20.1	33.5	6.7	33.4	TDS
19.8	66	29.7	3.3	53	23.1	19.8	287.1	330	89.1	198	26.4	19.8	3.3	13.2	16.5	224.4	3.3	16.5	3.3	3.3	3.3	3.3	3.3	36.3	220	9.9	16.5	3.3	16.5	SS
40.3551	81.864	56.891	42.4162	60.3141	38.7321	42.451	720.72	315.229	78.136	172.743	45.776	34.895	32.19	26.603	26.15 <del>3</del> 4	154. <b>302</b>	39.235	44.906	30.892	36.895	29.83	25.363	29.895	49.272	26.4441	31.312	35.488	25.795	42.152	ES

## Table 2: Hydrochemical analysis results

61	60	59	58	57	56	55	54	53	52	51	50	49	48	47	46	45	4	43	42	41	40	39	38	37	36	35	34	33	32	31
Arogbo	Orereara Road	Eruna	Idogun	Asisa	Orioke	Ayetoro	Obenla	Rewoye	Odonla	Awoye	Molutehin	Ilubirin	Kiribo	Itebukunmi	Salawa	Oboro	Inikorogha	Igbotu	Perearama-Zion	Obinehin	Araromi-Seaside 220	Temidire	Ogooluwayo	Ebute-Ipare	Ileriayo	Agbabu	Erinje	Abigi	Ilusin	Arafen
100	68	212	474	1176	1013	196	412	505	326	687	312	541	50	10	220	10	30	201	50	350	220	190	90	300	60	1500	40	30	10	30
6.2	6.8	7.6	6.7	4.4	6.7	7.6	7.2	7.42	7.4	7.6	7.71	6.1	6.78	5.71	7.25	5.82	6.57	6.9	7.14	8.05	8.6	7.89	7.93	8.07	7.24	7.34	5.63	5.6	5.04	7.27
100	52	148	190	1647	1421	146	90	58	58	48	89	89	34	12	84	28	20	80	38	218	128	84	102	128	78	316	10	10	10	32
58	24	76	84	267	327	76	34	18	18	32	39	46	26	8	60	14	140	44	16	150	106	30	56	116	24	162	4	4	6	12
32	28	72	106	1380	1102	70	56	40	40	16	29	22	8	4	24	14	6	36	22	68	22	54	46	12	54	154	6	6	4	20
0.1	0.32	0.09			1.76				1.32		0.07		2.21	0	17.6	2.11	0.7	1.89	1.76	4.23	3.01	2.29	8.1	ND	3.56	6.6	2.98	5.12	3.78	4.84
103		539	1052	1500	1500	87	75	90	140	83	98	40	12.9	6.99	11.9	8.57	8	14	8	21	7	8	11	20	12	384	5.99	12	7.99	12
ND	0.001	0.002	0.003	0.002	0.003	0.002	0.003	0.002	0.003	0.002	0.001	0.001	ND	ND	0.022	0.021	0.01	0.01	0.014	0.019	0.015	0.005	ND	0.016	0.011	0.018	0.02	ND	0.013	0.028
40	50	258	330	30	105	305	130	120	166	205	260	32.5	66	16	54	22	12	64	14	76	30	12	14	40	22	200	4	8	8	8
32	0	0	1	2	2	0	1	0	0	0	1	0	Х	Х	Х	Х	Х	Х	29.7	53.2	44.7	18.6	29.7	45.3	38.2	41.7	54.2	19.8	37.4	23.8
0.15			0.05	0.07	0.09					0.02			0	0	0	0.69	0.01	0.08	14	76	30	12	14	40	22	0.67	0.67	0.55	0.45	0.39
18 X	17.5 5			1200 3			48.8 1			54 2	63.7 2		8.39 X	4.54 X	7.73 X	5.57 X	5.2 X	9.1 X	5.2 1		4.55 3	5.2 1	7.15 1	13 4	7.8 2	250 2	3.89 4	7.8 8	5.19 8	7.8 8
	50	58	30	0	05	05	30	20	66	05	60	2.5							14	76	30	12	14	40	22	200	-			
Х	9.62	30.5	33.7	134	131	30.5	13.6	7.21	16	12.8	15.6	18.4	10.4	3.21	24	5.61	5.61	17.6	6.41	60.1	42.5	12	22.4	46.5	9.62	64.9	1.6	1.6	2.4	4.81
Х	6.83	17.6	25.9	337	267	17.1	13.7	9.76	3.42	3.9	7.08	5.37	1.95	0	5.85	14	1.46	8.78	5.37	16.6	5.37	13.2	11.2	2.93	13.2	37.6	1.46	1.46	0.98	4.88
70	46	142	245	788	678	131	276	338	218	460	209	362	33.5	6.7	147	6.7	20	135	33.5	235	147	127	60.3	201	40.2	1005	26.8	20.1	6.7	20.1
ND	22	70	121	970	836	64.7	136	167	108	227	103	179	Х	Х	Х	Х	Х	Х	16.5	115	73	63	29.7	99	19.8	495	13.2	9.9	3.3	9.9
121	55.2922	913.705	1763.297	2808.54	2806.11	168.305	134.816	154.3401	243.96	147.368	174.336	80.904	ND	ND	ND	ND	ND	ND	31.7571	107.321	66.09	31.29	49.659	91.05	44.7822	705.334	35.566	30.006	31.954	34.4061

## Table 2: Hydrochemical analysis results (continued)

Note;

TDS = Total Dissolved Solid; X = Not Tested	CBR = Chloride-Bicarbonate Ratio;
ES = Equivalent salinity; ND = Not Detected	SS = Suspended Solid

S/N	PARAMETER	WHO STANDARDS							
		Highest Desirable	Maximum Permissible						
1	Odour	Unobjectionable	Unobjectionable						
2	Colour	3.0 TCU	15.0 TCU						
3	pH at 20 <sup>0</sup> C	7.0 - 8.9	6.5 – 9.5						
4	Turbidity	5.0 NTU	5.0 NTU						
5	Conductivity	900 (µs/cm)	1200 (µs/cm)						
6	Total Dissolved Solid (TDS)	500 mg/l	1500 mg/l						
7	Total Hardness	100 mg/l	500 mg/l						
8	Calcium Hardness CaCO <sub>3</sub>	Not Specified	Not Specified						
9	Magnesium Hardness	20 mg/l	20 mg/l						
10	Nitrate (NO <sub>3</sub> )	10 mg/l	50 mg/l						
11	Iron (Fe)	1 mg/l	3 mg/l						
12	Alkalinity	100 mg/l	100 mg/l						
13	Manganese (Mn)	0.1 mg/l	0.4 mg/l						
14	Calcium (Ca <sup>2+</sup> )	Not Specified	Not Specified						
15	Magnesium (Mg <sup>2+</sup> )	20 mg/l	20 mg/l						
16	Sulphate $(SO_4^{2-})$	250 mg/l	500 mg/l						
17	Chloride (Cl <sup>-</sup> )	200 mg/l	250 mg/l						
18	Sodium (Na)	200 mg/l	250 mg/l						
19	Bicarbonate (HCO <sub>3</sub> )	Not Specified	Not Specified						
20	Suspended Solids	Not Specified	Not Specified						
21	Aluminium (Al)	0.2 mg/l	0.2 mg/l						
22	Ammonia	Not Specified	Not Specified						

## Table 3: Water Physico-Chemical Standards (After WHO, 2008)



Figure 5: Chloride map

www.iiste.org

study area, where there are functioning boreholes from which water samples were taken. Perhaps if water samples were taken at the same depth as it was in southeastern part they may likely yield high salinity.

## 7. RECOMMENDATION

To resolve this ambiguity and to determine the depth to saline-fresh water interface, the lateral extent of saline water intrusion and depth to fresh water in the study area, surface geophysical methods such as vertical electrical sounding and time domain electromagnetic should be employed and every available borehole logs in the area should be analyzed in order to delineate the subsurface layers and their water content.



Figure 6: Alkalinity map



Figure 7: Equivalent Salinity map



Figure 8: Total dissolved map



#### REFERENCES

Billman, H.G. (1992). Offshore Stratigraphy and Paleontology of the DahomeyEmbayment, West African. Nigerian Association of Petroleum Explorationists Bulletin,7(2):121-130.

Burke, K.C., Dessauvagie. J.F.T., Whiteman. A.J. (1971). The Opening of the Gulf of Guinea and the Geological History of the Benue Depression and Niger Delta Nature. Physical Science, 233 (38):51-55.

Choudhury, K, Saha, D.K., and Chakraborty, P, (2001). Geophysical Study of Saline Water Intrusion in a Coastal Alluvial Terrain. Elsevier Journal of Applied Geophysics, (46):189-200.

Ghyben, B. W. (1888). Nota in verband met de voorgenomen putboring nabij Amsterdam. Tijdschrift Kon. Inst. Ing., 8-22.

Herzberg, A. (1901). Die Wasserverzorgung Einiger Nordseebaden. Z. Gasbeleucht. Wasserverzorg, (44):815-844.

Klemme, H.D. (1975). Geothermal Gradient Heat flow and hydrocarbon Recovery In:A.G. Fischer and Judson, S. (Eds). Petroleum and Global Tectonics. Princeton University Press. 251- 304.

Mohammad, V. (2014). Drainage, Waterlogging and Salinity. Archives of Agronomy and Soil Science, 60(2):1625-1640.
Omosuyi, G.O., Ojo, J.S. and Olorunfemi, M.O. (2008). Geoelectric Sounding to Delineate Shallow Aquifers in the Coastal Plain sands of Okitipupa Area, Southwestern Nigeria. The Pacific Journal of Science and Technology. 9(2):562-577.
Omoyoloye, N.A., Oladapo, M.I. and Adeoye, O.O. (2008). Engineering Geophysical Study of Adagbakuja Newtown Development SW, Nigeria. Medwell Online Journal of

Earth Sciences. 2 (2): 55 - 63.

Petroleum Trust Fund (1997). Geological Map of Ondo State: National Rural Water Supply Project.

Rahaman MM and Bhattacharya AK (2014). Saline water Intrusion in Coastal Aquifer: A

case study from Bangladesh. Journal of Engineering. 4(1):7-13.

Whiteman AJ (1982). Nigeria: its Geology, Resources and Potential. Graham and

Trottam, London, 394pp.

World Health Organization (2008). Water and Sanitation. Guidelines for Drinking Water

Quality, 3<sup>rd</sup> Edition. Geneva, Switzerland.

www.who.int/water sanitation health/GDWQ. 668p.

The IISTE is a pioneer in the Open-Access hosting service and academic event management. The aim of the firm is Accelerating Global Knowledge Sharing.

More information about the firm can be found on the homepage: <u>http://www.iiste.org</u>

## **CALL FOR JOURNAL PAPERS**

There are more than 30 peer-reviewed academic journals hosted under the hosting platform.

**Prospective authors of journals can find the submission instruction on the following page:** <u>http://www.iiste.org/journals/</u> All the journals articles are available online to the readers all over the world without financial, legal, or technical barriers other than those inseparable from gaining access to the internet itself. Paper version of the journals is also available upon request of readers and authors.

### **MORE RESOURCES**

Book publication information: http://www.iiste.org/book/

Academic conference: http://www.iiste.org/conference/upcoming-conferences-call-for-paper/

## **IISTE Knowledge Sharing Partners**

EBSCO, Index Copernicus, Ulrich's Periodicals Directory, JournalTOCS, PKP Open Archives Harvester, Bielefeld Academic Search Engine, Elektronische Zeitschriftenbibliothek EZB, Open J-Gate, OCLC WorldCat, Universe Digtial Library, NewJour, Google Scholar

