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Isolation and Structural Elucidation of an Alkaloid Constituent from the Berries of Brucea Antidysenterica

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Abstract

A compound was extracted from the berries of *Brucea antidysenterica* and identified as Q4 (12-(methylamino)tridecane-6-ol). Its structure determination was based on ¹H, ¹³C NMR, DEPT-135, IR and LC-MS spectral measurements as well as comparison with the literature data. **Keywords:** *Brucea antidysenterica*; NMR; MeOH; Structure

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Introduction

Brucea is a genus of plant in the family Simaroubaceae. It is named for the Scottish scholar and explorer James Bruce [1]. *Brucea* is a genus of about 10 species occurring in Africa, Asia and Australia [3]. In Ethiopia two species are found and *Brucea antidysenterica* is the most well known of these. It is shrub or small tree from 10 to 15 m high, with smooth bark of grey to pale brown color with distribution in tropical Africa: Guinea, Nigeria, Ethiopia, Cameroon, Congo (DRC), Burundi, Sudan, Angola, Zambia and Malawi. The young stems are covered with ferruginous pubescence, leaves are 10-64 cm long and seed 8-9 mm long and 5-6 mm wide. Up-land (1400 to 2800 m) high evergreen forest and forest margins are very suitable area for the plant [2]. The plant is present at various geographical locations and their presumptive folklore used to prescribe for ascaris and diarrhea. In this paper 1D NMR, IR and LC-MS techniques were used to assign the NMR signals of the isolated sample, including ¹H NMR, ¹³C NMR, DEPT-135, IR and LC-MS spectra.

Materials and Methods

Plant Collection and Identification

The berries of *Brucea antidysenterica* were collected from southern nations, nationalities, and peoples' region of Ethiopia from Humbo woreda, which is around 337.5 km, south of Addis Ababa. The plant was identified by botanist in the department of biology, Wolaita Sodo University.

Instruments

 1 H NMR (400 MHz , CDCl₃) and 13 C NMR (100 MHz , CDCl₃) spectra were recorded on a Bruker ARX- NMR spectra and with TMS as an internal standard (chemical shifts in δ , ppm). The isolated compound was dissolved in CDCl₃ and analyzed with one-dimensional NMR (proton 1 H, carbon 13 C). ESI- MASS/MS spectra were recorded on an LC-MS mass spectrometer.

Extraction and isolation

The air-dried berries of *Brucea antidysenterica* (200 g) were extracted with one liter of methanol. The MeOH soluble crude (9 g) was chromatographed on silica gel (180 g) using gradient elution with *n*-hexane-EtOAc (100:1 to 1:1). Compound Q4 (35 mg) dark powder was obtained from the *n*-hexane-EtOAc (1:1) fraction.

Results and discussion

Compound Q4 was obtained as a dark powder from MeOH extract. Its molecular formula, $C_{14}H_{31}NO$ was determined by negative LC-MS. In the negative LC-MS spectrum, the quasi-molecular ion peak was at m/z 229.65 [M-H]⁻.

In the IR (KBr) spectrum the absorption band at 3435 cm⁻¹ due to secondary amine (R_2NH) and hydroxyl group. Strong absorption band at 2922 cm⁻¹ and medium absorption band at 1442 cm⁻¹ due to saturated C-H stretching.

The ¹H NMR spectrum (table 1) exhibited signal for the presence of a methine bearing of alcohol at $\delta_{\rm H}3.21(1\text{H}, \text{tt},)$. Signal at $\delta_{\rm H}2.79(1,\text{m})$ shows methine bearing of amine group. Sharp singlet peak at $\delta_{\rm H}2.47$ (3H) is due to N-methyl proton. Signal at $\delta_{\rm H}2.55(1\text{H})$ shows proton of alcohol and signal at $\delta_{\rm H}1.90(1\text{H})$ shows proton of nitrogen. Signals at $\delta_{\rm H}1.29-1.44(18\text{H}, \text{m})$ are due to methylene protons. Signal at $\delta_{\rm H}1.10(3\text{H},\text{d})$ shows methyl proton attached to carbon bearing amine group and signal $\delta_{\rm H}0.96(3\text{H},\text{t})$ is due to methyl.

In the ¹³C NMR spectrum, there were fourteen carbon signals. Signal at $\delta_C 71.9$ represent alcohol bearing methine carbon and signal at $\delta_C 54.7$ shows methine carbon bearing amine group and signal at $\delta_C 34$ is methyl substituent attached to nitrogen. Signals at $\delta_C 22.7$, 32.15, 23, 37.9, 23.5, 30, 24.67 and 37.77 are due to aliphatic

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carbons. Signals at $\delta_C 14.2$ and 21.3 are shows methyl carbons.

The multiplicity of each carbon atom was determined using DEPT-135 experiment, which revealed the presence of three methyl groups (one is attached to nitrogen, one is attached to aliphatic carbon and the other one is attached to methine carbon). Two methine group (one bearing the amine and the other one bearing the alcohol) and the other nine are aliphatic carbons. (Table 1)

Position	¹ H-NMR (ppm)	¹³ C-NMR(ppm)	DEPT-135
1	0.98(3H,t)	14.2	CH_3
2	1.34(2H, tt)	22.7	CH_2
3	1.29(2H, tt)	32.15	CH_2
4	1.29(2H, tt J= 5.98 Hz)	23	CH_2
5	1.44(2H, dt)	37.9	CH ₂
6	3.21(1H, tt)	72	СН
7	1.44(2H, dt)	37.9	CH ₂
8	1.29(2H, tt)	23.5	CH ₂
9	1.29(2H, tt)	30	CH ₂
10	1.29(2H, tt)	24.67	CH ₂
11	1.4(2H, dt, J=4.4Hz)	37.77	CH_2
12	2.79(1H,m)	54.7	СН
13	1.1(3H, dt)	21.3	CH ₃
N-CH ₃	1.00(3H,tt)	34	CH ₃
N-H	1.90(1H)	-	-
O-H	2.55(1H)	-	-

Table 1. ¹H-NMR, ¹³C-NMR and DEPT-135 spectral data of Q₄ in CDCl₃

All chemical shift data in the 1D-NMR, IR and LC-MS spectra closely matched the tentative proposed structure for the compound Q4(12-(methylamino)tridecane-6-ol). (See Fig 1)



Fig. 1 The structure of compound Q4

Conclusion

This work resulted in the isolation of one new compound that is not isolated from the plant in previous study, the compound was Q4 (12-(methylamino)tridecane-6-ol). The structure of the compound was characterized on the basis of spectral data (1H-NMR, ¹³C-NMR, DEPT-135, IR and LC-MS) as well as comparison with the literature data.

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